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PAOLA ESPOSITO, MARCELLO NICOLETTI

**Isolation of Monotropein, an iridoid glucoside from
*Arbutus unedo***

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SEZIONE II

(Fisica, chimica, geologia, paleontologia e mineralogia)

Chimica organica. — *Isolation of Monotropein, an iridoid glucoside from Arbutus unedo.* Nota di PAOLA ESPOSITO^(*) e MARCELLO NICOLETTI^(**), presentata^(***) dal Corrisp. G. B. MARINI-BETTÒLO.

RIASSUNTO. — Dalle foglie di *Arbutus unedo* (Ericaceae) è stato isolato un glucoside iridoide, la monotropeina *1*, identificata in base ai dati spettroscopici dello stesso iridoide e dei suoi derivati. Viene inoltre riportato l'uso di LiAlH_4 quale agente riducente stereospecifico di anelli epossidici in glucosidi iridoidei.

In this paper we report the isolation and identification of an iridoid glucoside, *1*, isolated from the leaves of *Arbutus unedo* (Ericaceae). In the same plant another iridoid glucoside, unedoside *2*, containing an epoxidic ring is present [1].

1 is a crystalline compound (m.p. 163–5 °C, from Me OH–H₂O) showing acidic properties it has a molecular formula $\text{C}_{16}\text{H}_{22}\text{O}_{11}$ and Rf 0.21 (bluish-violet spot with vanillin reagent). *1* shows UV absorption at 235 nm ($\lg \epsilon = 3.9$), characteristic of an iridoidic enol-ether system, conjugated with a carbonyl group [2], IR bands at 1700 and 1650 cm^{-1} and in the mass spectrum peaks at m/e 359 ($\text{M}^+ - 31$) and 331 ($\text{M}^+ - 59$) are present. Like other naturally occurring iridoid glucosides, *1* is converted by acid hydrolysis into glucose (1 mole) and insoluble black products, due to the decomposition of the aglycone.

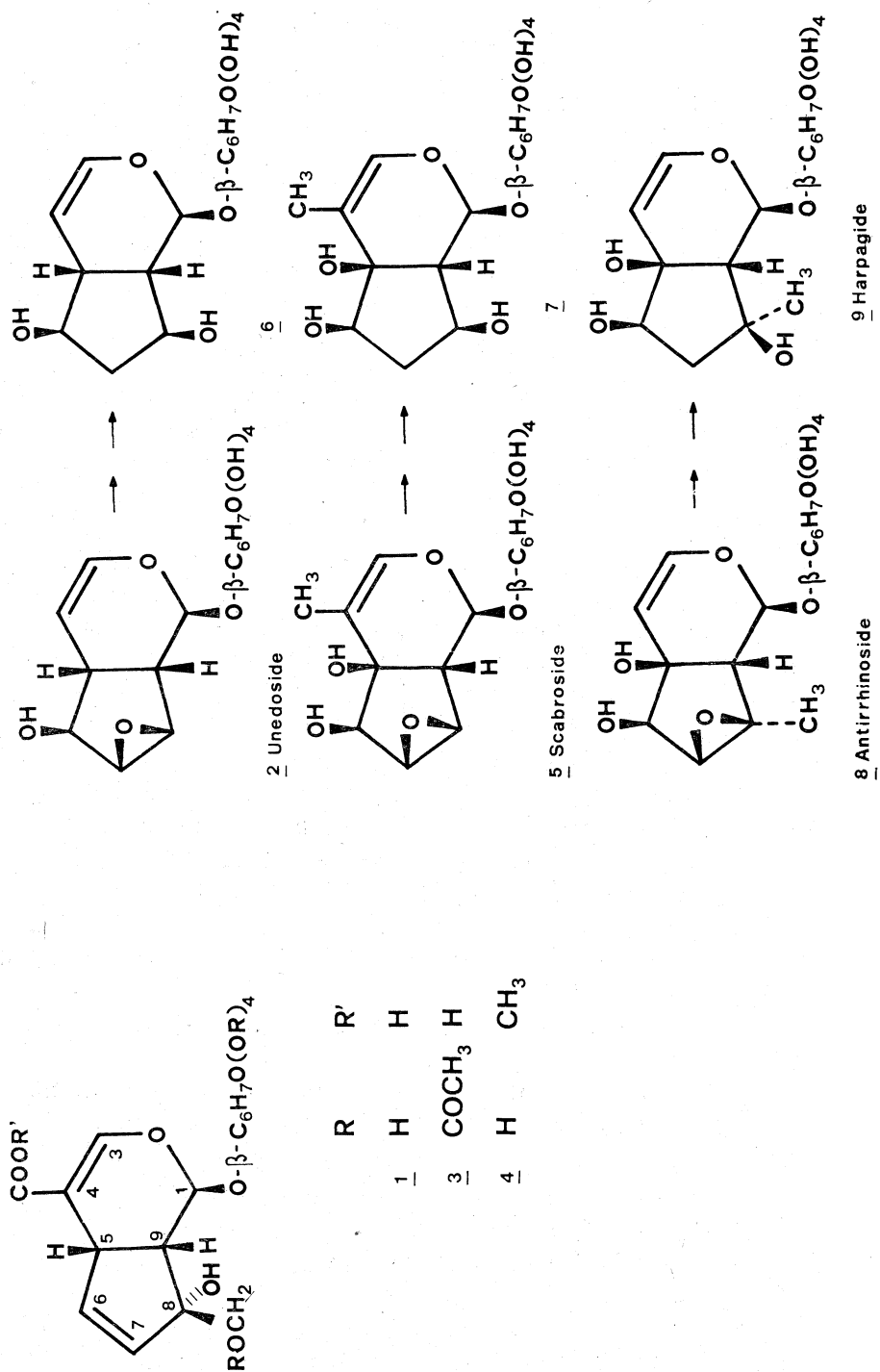
The ^1H NMR spectrum of *1* in D_2O confirms the iridoid structure and, in particular, shows: a doublet at δ 7.39 ($J = 1.0$ Hz), attributable to H–3, shifted to lower fields by the presence of the carbonyl group at C (4), in accordance with previous UV and IR data; two double doublets at δ 6.21 ($J = 6.0$ and 2.4 Hz) and 5.68 ($J = 6.0$ and 1.8 Hz), attributable to two vicinal olefinic protons in the cyclopentane ring; a doublet at δ 5.60 ($J = 2.0$ Hz), assignable to H–1; a double doublet at δ 2.66 ($J = 8.0$ and 2.0 Hz) due to H–9.

Acetylation of *1* under mild conditions afforded the pentaacetate *3* ($\text{C}_{26}\text{H}_{32}\text{O}_{16}$), m.p. 176–8 °C, which shows in the IR spectrum a residual OH band, which could be attributed to a tertiary hydroxyl group. The ^1H NMR spectrum of *3* (CDCl_3), in addition to the signals analogous to the corresponding ones in *1*, evidences the presence of five acetoxy groups (four belonging to the glucose and one to the aglycone).

(*) Istituto di Chimica Organica, Università di Roma, p.le Aldo Moro, 5, Roma.

(**) Cattedra di Chimica Generale ed Inorganica, Università di Roma, p.le Aldo Moro, 5, Roma.

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Finally **1** was converted by reaction with CH_3N_2 into the methylester derivative **4**, which shows in the ^1H NMR spectrum (D_2O) the appearance of the signal at δ 3.70, correspondent to the carbomethoxy group. Spectroscopic evidences were corroborated by detailed spin decoupling experiments carried out on compound **4** (in methanol- d_4). In fact the irradiation at δ 3.8, multiplet relative to H-5, partially masked by glucose signals, converted the double doublets H-6, H-7 and H-9 into doublets ($J = 6.0, 6.0$ and 2.0 Hz, respectively); irradiation of the doublet at δ 5.58 (H-1) simplified the double doublet assigned to H-9 into a doublet ($J = 8.0$ Hz); finally irradiation at δ 2.70 (H-9) transformed H-1 into a sharp singlet and generated some changes into the overlapped zone between δ 3.6-4.0.

Spectroscopical and physical data of **1** are in agreement with those reported for monotropein [4].

In other plants [5] compounds having C(7, 8) β -epoxide function are coupled with compounds having C(8) β -hydroxyl function. Usually these substances have been chemically correlated by reductive cleavage of the epoxide function with Li in liquid NH_3 [5]. Disadvantages of this reaction are the low temperature (-70°C) and the difficulty in obtaining a good medium for the reaction.

We investigated an alternative and more convenient method. Good results were obtained using LiAlH_4 on the O-acetyl derivatives in THF at 25°C for 2.5 h. The reaction was achieved on unedoside **2** and scabroside **5** obtaining as the only product the corresponding C(8) β -hydroxyl iridoid compounds **6** and **7** respectively. Similarly antirrhinoside **8** gave harpagide **9**, but to raise the yield of the reaction it was necessary to operate at 70°C , in accordance with the lower reactivity of substituted epoxide function [6].

REFERENCES

- [1] H. RIMPLER and H. PISTOR (1974) - «Z. Naturforsch.», 29 c, 368.
- [2] a) A. BIANCO, C. BONINI, M. GUIZO, C. IAVARONE, and C. TROGOLO (1977) - «Gazz. Chim. Ital.», 107, 67. b) M. L. SCARPATI and M. GUIZO (1969) - «Gazz. Chim Ital.», 99, 1150.
- [3] J. M. BOBBITT and K. P. SEGEBARTH (1969) - in «Cyclopentanoid terpene derivatives», W. I. Taylor and A. R. Battersby eds; M. Dekker, N. Y., p. 70, 76.
- [4] Ref. 3, p. 61.
- [5] a) S. R. JENSEN, B. J. NIELSEN and R. DAHLGREN (1975) - Bot. Notiser, 128, 148. b) P. ESPOSITO, M. GUIZO, M. NICOLETTI and C. DE LUCA (1976) - «Gazz. Chim. Ital.», 106, 57.
- [6] E. L. ELIEL and M. N. REVICH (1960) - «J. Org. Chem.», 83, 1362.