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ELISABETTA FORESTI, LODOVICO RIVA DI SANSEVERINO

**The x-ray crystal and molecular structure of an  
organic mineral: simonellite, C<sub>19</sub>H<sub>24</sub>**

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**Mineralogia.** — *The x-ray crystal and molecular structure of an organic mineral: simonellite, C<sub>19</sub>H<sub>24</sub>* (\*). Nota di ELISABETTA FORESTI e LODOVICO RIVA DI SANSEVERINO (\*\*), presentata (\*\*\*), dal Socio P. GALLITELLI.

**RIASSUNTO.** — Si descrive uno studio cristallografico della simonellite, chimicamente 1,1-dimetil-7-isopropil-1,2,3,4-tetraidrofenantrene. I dati cristallografici sono  $a = 9.231 \text{ \AA}$ ,  $b = 9.134 \text{ \AA}$ ,  $c = 36.01 \text{ \AA}$ ,  $Pnna$ ,  $Z = 8$ .

Mentre ogni tentativo di risolvere la struttura per mezzo di un calcolo tridimensionale Patterson risultava inutile, l'impiego dei metodi diretti attraverso la formula «sigma 2» di Karle e Karle (1966) ha rivelato subito 18 dei 19 atomi di carbonio della simonellite.

Il raffinamento con il metodo dei minimi quadrati con fattori di temperatura isotropi ha dato un  $R$  finale di 12%, su 2432 riflessi misurati.

Si discute la stereochemica intramolecolare e si prospetta una possibilità di biogenesi per questo minerale organico, trovato come incrostazioni biancastre in una miniera di lignite di Fognano, Siena.

#### INTRODUCTION.

Simonellite had been studied in this Institute by Boeris (1919) about fifty years ago; it appears as white-yellowish crusts in the lignite mine, at Fognano, Siena (Italy).

Optical, morphological and crystal data were then collected by Emiliani and Pellizzer (1952, 1953) and *x-ray* crystal structure analysis began during the fall 1964. Ghigi and Fabbri (1965) studied simonellite by chemical and spectroscopic means and then synthetized the mineral (Ghigi *et al.*, 1968). This is a report of the crystal structure analysis.

#### EXPERIMENTAL.

Simonellite, C<sub>19</sub>H<sub>24</sub>, M.W. 252, M.P. 59°–60° C; samples selected in the Museum of this Institute and recrystallized from benzene. Orthorhombic, *Pnna* (1) from systematic absences; cell dimensions measured by Emiliani and Pellizzer (1953) have been confirmed,  $a = 9.231 \pm 0.003 \text{ \AA}$ ,  $b = 9.134 \pm 0.003 \text{ \AA}$ ,  $c = 36.01 \pm 0.01 \text{ \AA}$ . They have been calculated by a least squares refinement on 84 reflection angles  $\theta$ , obtained from  $o\ k\ l$  and  $h\ o\ l$  Weissenberg photographs, calibrated with Al (Van den Berg, 1964),  $V = 3037 \text{ \AA}^3$ ,  $D_m = 1.08$ ,

(\*) Istituto di Mineralogia e Petrografia dell'Università di Bologna.

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(\*\*\*) Nella seduta del 12 giugno 1969.

(1) This space group has been preferred to the conventional *Pccn*, to respect a tendency to tetragonal symmetry pointed out by Emiliani and Pellizzer (1953).

$Dx = 1.10$ ,  $Z = 8$ . Crystal dimensions  $0.5 \times 0.5 \times 0.6$  mm., radiation used CuK ( $\lambda = 1.5418 \text{ \AA}$ ).

Of the 2432 densitometrically measured reflections, 899 have been classified as "unobserved", giving them  $1/2$  of the minimum observed intensity. Data have been recorded with the Weissenberg multiple film equiinclination technique along the  $a$  axis ( $0 \leq k \leq 7$ ) and corrected for spot deformation (Riva di Sanseverino, 1967) but not for absorption. Scaling was assured through cross layer correlation calculations.

Atomic scattering factors have been taken from the International Tables for Crystallography, vol. III. During the first steps computer programs written by Albano *et al.* (1963) have been used, but more recently the Stewart "X Ray 63" system (1964) has been adopted.

#### THE FIRST TRIAL FOR STRUCTURE SOLUTION.

As the simonellite structure (fig. 1) was suspected to be similar to retene (Ghigi and Fabbri, 1965), it was then supposed that the stereographic projection of the Patterson function could reveal the molecular plane. This method,

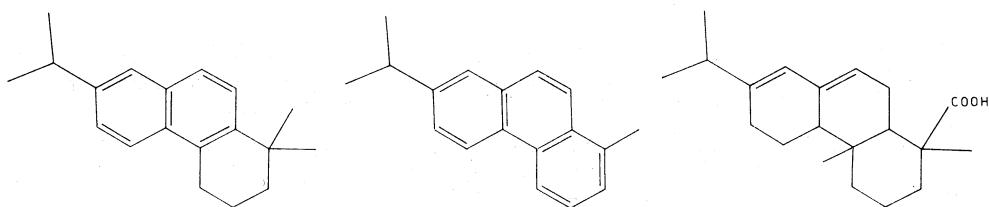


Fig. 1. - Simonellite (a) and the two related structures, retene (b) and abietic acid (c).

developed by the MacGillavry school in Amsterdam (Stam and Riva di Sanseverino, 1966; Bart and MacGillavry, 1968), consists in building spheres of a chosen radius around the origin of the Patterson function and then in plotting a stereographic projection of the maxima lying on the sphere.

In the case of planar hydrocarbon molecules, containing chains or rings, typical sequences of 3 peaks will be aligned along the same major circle in the stereographic projection of the  $1.4$  and the  $2.4 \text{ \AA}$  spheres. Alternatively three  $1.4 \text{ \AA}$  peaks will be disposed at  $60^\circ$  interval, as well as the three  $2.4 \text{ \AA}$  peaks, as shown by the scheme in fig. 2c. Peak heights will be proportional to the vector multiplicity in the internal geometry of the molecule.

Fig. 2a more than fig. 2b shows these features. The direction cosines of the normal (full square in fig. 2) to the supposed molecular plane in the  $x, y, z$  coordinates system are respectively  $\cos \varphi = 0.54$ ,  $\cos \chi = 0.78$ ,  $\cos \psi = 0.36$ .

Although these values were later confirmed (Table V) no trial succeeded in the positioning of the molecule. A noteworthy influence on these negative

approaches could possibly be caused by the high symmetry of the orthorhombic system and/or by the not perfect planarity of the molecule and/or by the exaggerate grid of the Patterson function, forcedly imposed by the contrasting conditions of large cell dimensions and of the relatively small IBM 1620 computer.

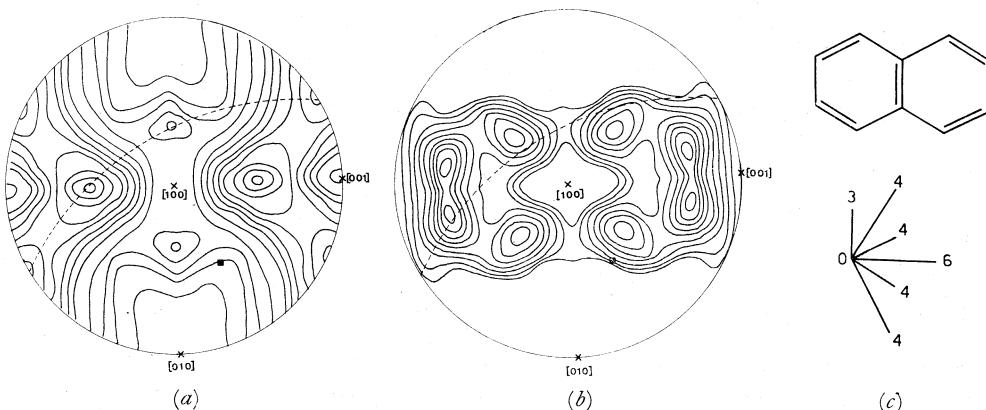


Fig. 2. — Stereographic projection of spherical sections around the origin of the Patterson function for simonellite with radii 1.4 Å (a) and 2.4 Å (b). The dashed lines indicate the probable plane of two of the eight molecules in the unit cell and the full square the emergence of the normal to the plane. As an example, naphtalene and its Patterson vectors pattern are reported in (c): numbers indicate peak multiplicity.

#### THE APPLICATION OF $\Sigma_2$ .

More recently a bigger computer (IBM 7094) and the development of direct methods (see Karle and Karle, 1966) encouraged a new attack to the structure solution from a different point.

As only the listing of E's and the relative statistics were available, a program for the application of the  $\Sigma_2$

$$sE_h \cong s \sum_k E_h E_{h-k}$$

was prepared (Foresti and Guerrieri, 1969). While the statistics (Table I) confirmed the centrosymmetric space group, the criteria for the choice of the origin (Stout and Jensen, 1968) were applied, giving positive signs to reflections 4, 5, 16; 1, 1, 31; and 4, 4, 25.

These assignments were sufficient to attribute signs to 134 of the 153 chosen E's with  $E > 1.85$ , probably owing to the large number of internal relations among orthorhombic structure factors. When the refinement was terminated, it was found that only three signs were inverted.

The 134 signs produced the first E map, which showed all atoms but one (C16); this last was revealed by a successive three-dimensional  $F_o$  Fourier, including all the observed structure factors.

A full matrix least squares refinement cycle with isotropic temperature factors lowered R from 28 % to 18.8 %. The successive difference fourier gave significant positive electron density at 20 sites which were correspondent to good positions for hydrogen atoms. The remaining 4 were assigned theoretical positions and the structure factor calculation gave R = 16.7 %.

TABLE I.  
*Normalized structure factors.*

Values	Experimental for Simonellite	Theoretical for	
		Centrosymmetric	Non-centrosymmetric
$\langle  E  \rangle$ . . . . .	0.745	0.798	0.886
$\langle  E^2 - 1  \rangle$ . . . . .	1.083	0.968	0.736
$\langle  E ^2 \rangle$ . . . . .	1.000	1.000	1.000
$E > 1$ . . . . .	30.5%	32.0%	36.8%
$E > 2$ . . . . .	5.0%	5.0%	1.8%
$E > 3$ . . . . .	0.9%	0.3%	0.01%

Two further refinement cycles as before, with hydrogen coordinates held constant ( $B = 6 \text{ \AA}^2$ ), brought R (2) to 12.0 % and refinement was terminated, as standard deviations and stereochemical results seemed satisfactory.

Final parameters are listed in Table II, observed and calculated structure factors in Table III (3), bond distances in fig. 3 and angles in Table IV. Standard deviations for distances, obtained from the final least squares refinement cycle are  $\bar{\sigma} = 0.008 \text{ \AA}$  for C—H ( $sp^2$ ) with a maximum of 0.009  $\text{\AA}$  for C<sub>9</sub>—C<sub>10</sub> and  $\bar{\sigma} = 0.009 \text{ \AA}$  for C—H ( $sp^3$ ) with a maximum of 0.01  $\text{\AA}$  for C<sub>17</sub>—C<sub>18</sub>.

#### DISCUSSION.

It is encouraging to notice from Table II that the refined temperature factors are highest for the atoms departing from the phenanthrene nucleus, as expected. Furthermore, while atom C<sub>16</sub> could not be found, atoms C<sub>15</sub>, C<sub>17</sub>, C<sub>18</sub> and C<sub>19</sub> appeared well as peaks in the E map, but with a height about 30 % lower than the average peak height.

(2) The last cycle was run with weights, following the expression recommended by Cruickshank *et al.* (1961) for photographic data.

(3) Reflections 004, 006, 008, 019 and 0, 2, 10 were discarded as they were too strong to be measurable.

TABLE II  
*Fractional positional parameters with s.d. and temperature factors.*

ATOM	<i>x</i>	<i>y</i>	<i>z</i>	$\sigma(x)$	$\sigma(y)$	$\sigma(z)$	$B (\text{\AA}^2)$
C <sub>1</sub> . . . .	0.09458	0.58012	0.18070	0.00066	0.00059	0.00015	3.45
C <sub>2</sub> . . . .	0.11495	0.47394	0.21374	0.00071	0.00063	0.00015	4.07
C <sub>3</sub> . . . .	0.26874	0.46425	0.22579	0.00072	0.00065	0.00016	3.83
C <sub>4</sub> . . . .	0.36318	0.40037	0.19476	0.00067	0.00060	0.00015	3.47
C <sub>5</sub> . . . .	0.56490	0.37095	0.13186	0.00067	0.00061	0.00014	3.46
C <sub>6</sub> . . . .	0.66610	0.35811	0.10416	0.00070	0.00064	0.00016	4.06
C <sub>7</sub> . . . .	0.64867	0.43797	0.07029	0.00064	0.00058	0.00014	3.31
C <sub>8</sub> . . . .	0.53031	0.52239	0.06625	0.00067	0.00060	0.00015	3.39
C <sub>9</sub> . . . .	0.29930	0.62198	0.08954	0.00073	0.00067	0.00016	3.97
C <sub>10</sub> . . . .	0.19724	0.63518	0.11755	0.00069	0.00063	0.00015	3.55
C <sub>11</sub> . . . .	0.21548	0.56019	0.15204	0.00063	0.00057	0.00014	2.92
C <sub>12</sub> . . . .	0.33539	0.47711	0.15772	0.00060	0.00054	0.00013	2.74
C <sub>13</sub> . . . .	0.44191	0.45964	0.12869	0.00059	0.00052	0.00013	2.60
C <sub>14</sub> . . . .	0.42167	0.53550	0.09458	0.00062	0.00055	0.00013	2.98
C <sub>15</sub> . . . .	0.09385	0.74048	0.19488	0.00076	0.00074	0.00016	4.73
C <sub>16</sub> . . . .	-0.05379	0.54551	0.16356	0.00081	0.00077	0.00019	5.07
C <sub>17</sub> . . . .	0.76321	0.42646	0.03970	0.00071	0.00067	0.00016	4.34
C <sub>18</sub> . . . .	0.91250	0.46952	0.05351	0.00090	0.00085	0.00020	5.95
C <sub>19</sub> . . . .	0.76532	0.27374	0.02213	0.00083	0.00084	0.00019	5.71

The average distances C—C ( $sp^2$ ) = 1.401 Å and C—C ( $sp^3$ ) = 1.539 Å agree extremely well with those published by Sutton (1965). The observed trend in simonellite follows a certain regularity and corresponds with features found in phenanthrene (Trotter, 1963). Trotter compares theoretical and observed values for each bond length and concludes that for *f* bond and *g* bond (see fig. 3) there is no agreement. This is repeated in the simonelite structure; again, bond *f* is noteworthy longer and bond *g* somewhat shorter than foreseen. The first seems to show a smaller, and the second a larger aromatic character than it might be deduced by the resonance structures or by LCAO bond orders (Dictionary of values of molecular constants, 1955).

A fair analogy with these findings has been also found in different aromatic compounds, with a similar disposition of fused rings (Ferrier and Iball, 1963 a, b).

Regarding bond angles (Table IV), simonellite shows a remarkable planarity at junctions, while angles at C5 and C8 are slightly greater than expected; tetrahedral angles do not reveal any particular strain. The average standard deviation for angles is  $0.5^\circ$ .

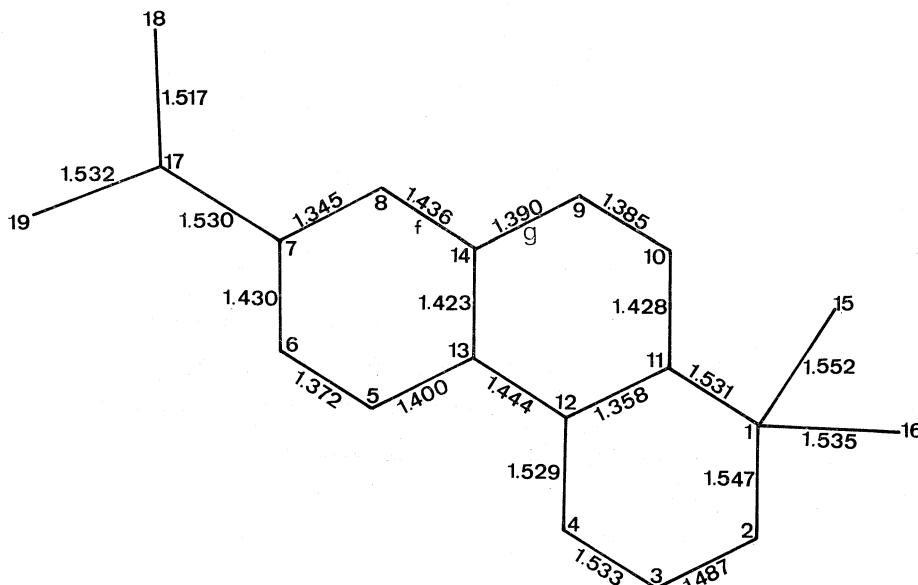


Fig. 3. - Numbering and bond distances in Å for simonellite.

The planarity of the aromatic part of the molecule is calculated through the equations ( $1\text{ D}$  and  $1\text{ A}$ ) listed in Table V, together with atomic distances in Angstrom from the least square plane. Two further equations ( $2\text{ D}$  and  $2\text{ A}$ ) are quoted for a comparison, calculated for the least square plane through the whole molecule. The equations have been listed both for fractional coordinates (direct space) and for actual position in the crystal cell (Angstrom space).

As already noticed, these equations (specially  $1\text{ A}$ ) correspond almost perfectly with that resulting from the stereographic projection of the Patterson function (fig. 2), regardless of the distance from the origin, obviously indeterminated in the three-dimensional synthesis.

The inverse and probably significant displacement of C4 and C5 needs one more comment; the same behaviour is shown by the equivalent atoms in the *1,2-cyclo-penteno-phenanthrene* molecule studied by Entwistle and Iball (1961). In the simonellite molecule a noticeable agreement is found with the considerations made by Trotter (1963) who attributed the effect to «slight intermolecular overcrowding involving the hydrogen atoms bonded to C4 and C5».

TABLE III.

Structure factors for simonellite: columns are  $l$ ,  $\text{IO F}_0$  and  $\text{IO F}_c$ .  
Unobserved reflections are marked with L.

	0+0,L	28	131 -123	6	29L 49	1,3,L	17	152 123	9	37L -47	2,1,L	18	103	73	
10	185	221	32	134 147	10	28L 37	1	117 82	19	35L -17	11	109 -136	0	343 392	
12	537	-482	34	64 -77	12	27L 17	2	239 223	20	36L 61	12	37L 71	1	96 95	
14	199	113	36	30L -52	14	26L -37	3	223 215	21	84 -75	13	129 148	2	512 -573	
16	222	-243	38	27L 27			4	619 -639	22	147 -129	14	37L -70	3	206 -208	
18	106	10	40	24L 26		0,11,L	5	488 -495	23	37L -25	15	37L -78	4	407 431	
20	100	-5					6	233 -162	24	37L 4	16	37L 44	5	593 686	
22	171	-181	0,5,L	1	88 -127	7	74 -57	25	83 -55	17	37L -17	6	696 764		
24	195	-140		3	89 155	8	22L -3	26	37L -24	18	36L -23	7	719 763		
26	129	112	1	103 78	5	138 175	9	23L -6	27	72 62	19	36L -40	8	539 -530	
28	313	-291	3	137 -96	7	57 -93	10	446 440	28	37L -11	20	35L -25	9	15L 4	
30	169	142	5	516 -583	9	63 -86	11	126 131	29	37L 14	21	119 -119	10	201 195	
32	421	398	7	26L 67	11	19L -22	12	212 -241	30	66 112	22	103 110	11	275 212	
34	245	-235	9	411 -394	13	17L 54	13	166 152	31	36L 10	23	33L 66	12	391 -299	
36	207	-197	11	170 -205	15	15L -22	14	585 526	32	172 151	28	32L 54	13	444 418	
38	88	-32	13	335 379			15	209 155	33	123 108	25	31L -6	14	276 -195	
40	64	30	15	68 -60	1,1,L	16	28L 5	34	83 69	26	73 -95	15	40 13		
42	57	-34	19	320 -317			17	78 79	35	60 -45	27	29L 29	16	473 440	
44	119	110	21	221 239	1	661 683	18	279 -262	36	59 -74	28	27L -17	17	115 -98	
	23	239	212	2	936 933	19	163 -163	37	79 -93			18	151 -145	38	
0+1,L	25	71	-69	6 364	-397	20	128 109	38	26L -13		1,9,L	19	23L 13	39	
27	113	-142	7	74 -110	21	123 -93	39	24L 30		20	53 67	40	44 40		
3	450	-403	29	34L 73	8 822	-757	22	63 66	40	55 43	1	37L -0	-21	88 92	
5	806	-810	31	87 91	9 554	524	23	183 -144	41	18L -41	2	37L 33	22	87 -89	
11	480	-402	35	29L 15	11 139	71 25	35L 40			3	37L -52	23	61 55		
13	213	214		12 121	-80	26	141 -133			4	66 65	24	62 -61		
15	193	-169	0,6,L	13 133	99 27	36L 35	1	71 51	6	95 75	26	300 255	0	443 426	
17	152	-149	14	597 542	28 37L	-43	2	93 81	7	36L 39	27	101 -104	1	155 138	
19	67	-85	0	257 -250	15 562	524	29	37L 0	3	311 -337	8	112 -95	28	127 -136	
21	26L	19	2 90	72 16	106 75	30	101 -76	4	269 253	9	103 -83	29	60 -10		
23	28L	-33	4 229	-203	17 200	-147	31 157	154	5 200	-190	10	125 -120	30	101 -97	
25	30L	8	6 335	-355	18 289	-245	32 93	6	78 104	11	35L 8	31 92	97 5		
27	224	216	8 350	336	19 430	-368	33 76	81	7 133	-13	126 35L	58 32	31L 23		
29	101	88	10 30L	-58	20 61	-34	34 36L	62	8 233 -249		33 31L	36 7	287 -299		
31	210	-203	12 31L	0	21 117	101 35	36L -51	9	318 358		1,10,L	34 56	46 8		
33	129	-121	11 126	146	22 70	54 36	161 -156	10	93 115		35 112	-87	9 535 527		
35	217	195	16 130	-128	23 144	-126	37 165	163	11 83	-91	1 33L 7	36 65	55 10		
37	143	145	18 353	338	24 32L	28 38	90 -80	12	71 71	2	32L 13	37 28L	-19 11		
39	30L	-33	20 34L	-10	25 61	2 39	30L -18	13 35L	61 3	74 -78	38 61	-63 12	23L 28		
41	73	-71	22 70	-76	26 115	116 40	134 133	14	105 -152	4	94 98	39 26L	-36 13		
43	23L	151	24 34L	48 27	185 -166	41 26L	20 15	138 -124	5	121 140	40 25L	29 14	158 -165		
45	59	64	26 169	-156	28 286	269 42	24L 29	16	36L 2	6	129 152	15 182	-185		
	28	33L	-17	29 137	114 43	52 42	17 130	126	7 32L -7		2,2,L	16 152	134 285 -244		
0+2,L	30	31L 19	10 30	429 -374	44 50	-58	18 63	-74	8 95	-94	9	139 122	27 31L	-32	
2	34L	-333	0,7,L	32 192	-195	1,4,L	20 37L	48 9	141 -133	0	548 520	18 84	-68		
4	730	-693	33 93	-83			21 113	-132	11 105	-104	2 744 784	20 165	-158		
6	529	529	1 176	-189	34 101	100 1	173 -179	22 106	120	12 29L 2	3 363 427	21 53	-50		
8	779	-790	3 130	117	35 150	-149	238 -230	23 189	189 13	100 113	4 282 -307	22 29L	-41		
12	322	306	5 32L	-38	36 95	78 3	266 -291	24 138	138 14	128 28L	4 5 539 -508	23 29L	53		
14	677	655	7 32L	48	37 137	-119	4 637	612 25	37L 24	15 27L -13	6 209 -156	24 86	88		
16	413	-355	9 33L	-38	38 34L	-27	5 196	211 26	110 -101		7 83 133	25 102	89		
18	581	521	11 33L	-63	39 35L	1 6	134 155	27 36L -39	1,11,L	8 88	-34	26 30L	8		
20	109	-78	13 194	202	40 31L	37	7 102	89 28	96 -86	9	139 122	27 31L	-32		
22	115	91	15 34L	5 41	291 16	8 426	-384	29 35L -58	1	51 -62	10 197	214	28 31L		
24	105	90	17 69	87 42	27L 11	9 648	664	30 34L	34 2	25L -32	11 106	100	29 31L	-44	
26	31L	-3	19 135	-137	43 25L	-16	10 209	197 31	68 56	3 24L 30	12 19L	4 30	30L 35		
28	33L	-11	21 109	140 44	22L -34	11 27L	-0	32 32L	-31 4	24L -3	13 230	225	31 30L	-18	
30	90	76	23 23L	33 26	45 79	101 12	28L -26	33 31L	-4 5	85 101	14 137	-160	32 86	-63	
32	324	323	25 82	-60	46 13L	68 13	28L -53	34 29L	8 6	24L 45	15 239	-204	33 29L	41	
34	131	-104	27 113	-90			14 29L -19	35 27L	12 7	23L -32	16 405	369	34 87	66	
36	238	-259	29 65	-66	1,2,L	15 30L	-5	36 35L	36 25L	36	17 240	-206	35 27L	-5	
38	30L	12	31 71	85			16 30L -30	37 32L	55 55	2,0,L	18 23L	-38	36 26L	-27	
40	28L	-1	33 23L	1	1 1419	1532	17 120	97 38	51 25	19 342	-346	37 140	-150		
42	73	-50	35 20L	22	2 1020	1094	18 323	-282	39 17L -7	4 615	-631	20 198	-191	38 108	
44	76	87	3 78	-14	19 33L	16 40	9 56	5 5	189 165	22 163	-128	40 66	-51		
0+3,L	0	34	59	6 583	585	22 159	156		6	189 165	22 163	-128	40 66	-51	
1	554	-573	2 34L	-20	7 162	-170	23 67	-42	1 137 121	9 381	-341	25 165	-127	43 10L	-64
3	479	408	4 42	-74	8 352	-324	24 129	133 2	35L 30	10 197	-166	26 413	-406		
5	542	-539	6 55	-101	9 152	-138	25 70	-26	3 155 -156	11 559	-474	27 377	360	2,5,L	
7	90	-83	8 111	114	10 122	-70	26 37L	-57	4 255 253	12 81	-22	28 30L	-41		
9	217	132	10 234	254	11 211	-170	27 186	-146	5 36L -0	13 409	435	29 340	-362		
11	508	-482	12 106	-110	12 190	-176	28 37L	-61	6 115 128	14 389	416	30 132	101		
13	1248	1176	14 64	-28	13 155	-210	29 108	-107	7 112 111	15 162	229	31 31L	-19		
15	345	350	16 33L	67	14 411	-363	30 27L	28 8	36L 14 16	16 923	894	32 31L	32 3 218		
17	298	-238	18 33L	-46	15 410	370	357 31	104 95	9 1 -91	17 391	-344	33 72	91	4 215 -187	
19	154	152	20 32L	40 16	65 -27	32 82	-73	10 80	82 18	76 -52	34 54	25 5	294 284		
21	49	49	52 22	31L	-19 17	27L 15	33 127	-125	11 82 70	19 544	-483	35 130	-126		
23	177	123	24 29L	13 18	4 16	-375	34 35L	-12	12 37L 3	21 439	427	37 28L	45 8 46		
25	32L	29	53	-64	19 29L	3 35	27 72	74	13 37L -21	24 37L	-22	37 28L	45 8 46		
27	107	-104	28 93	-109	20 57	67 51	6 29L -29	74	15 53 52	23 137	38	27L 34 9 137	-132		
29	117	114	30 22L	18	21 71	-37	13 142	15 23L -63	23 158 115	39 25L	7	10 119	-131		
31	353	-335	32 19L	-3	31 129	119	2 28L -55	25 66	61 -61	33 31L	29 3	194 -229	20 164		
33	275	-260			23 32L	-22	39 28L	-11	17 37L 12	24L 37L	41	24L 13	11 353		
35	241	-257	0,9,L	24	138	107 40	58 -76	18 37L	22 26	453 -473	42	89 80	13 170	148	
37	30L	-5	25 34L	23	41 23	25L 4	19 37L	-21	24 37L 3	21 439	427	14 83	-102		
39	28L	-12	1 94	-117	26 59	-4 42	20 37L	21 30	37L 66	28 29L -1	14	83 102			
41	78	-67	3 112	135	27 81	-105	21 66	60	29 431	-418	2,3,L	15 27L	45		
43	20L	-30	5 187	175	28 36L	2	1,5,L	22 157	-185	30 307	299	0	163 -131		
	7 33L	-2 29	37L -39		23 36L	2 31	36L 2	31 148	-139	1 100	-91	13 104	-102		
0+4,L	9														

TABLE III (continued).

2,5,L	24	65	-49	29	52	56	11	339	-308	35	123	-146	7	23L	-14	3	507	432	23	187	156							
33	76	54	26	24L	62	31	165	-139	13	67	-38	37	57	77	9	22L	-4	5	238	-243	25	387	347					
34	61	-60	27	52	6	32	30L	-25	14	118	89	38	111	-40	10	22L	-3	6	44	-13	26	34L	5					
35	25L	4	28	44	70	33	180	183	15	165	168					11	6L	72	7	54	19	27	34L	-35				
36	24L	15	29	20L	60	34	189	176	21	214	196		3,7,L			12	21L	-4	8	131	132	28	34L	28				
37	87	105	30	39	34	35	28L	-5	17	250	-210					13	21L	-7	9	21L	45	29	144	-126				
38	89	118	31	17L	-38	36	65	-59	18	88	75	1	29L	-5	14	49	8	10	191	-188	30	106	84					
39	56	-49	32	15L	0	37	71	-75	19	137	130	2	181	-193	15	59	47	11	63	85	31	32L	-15					
40	39	47					38	25L	-28	20	69	52	3	29L	51	16	16L	-75	12	54	-43	32	31L	-40				
							2,9,L	39	24L	-3	21	64	-73	4	29L	-76	18	63	57	13	24L	10						
																17L	-17	14	25L	-6			4,5,L					
																			26L	-9								
0	176	164	2	30L	25		3,2,L	23	29L	-1	6	49	-65	19	15L	-38	15	16	67	70	1	58	-51					
1	102	89	3	101	-69	1	307	252	25	325	300	8	29L	-34		4,0,L		17	137	132	2	51	42					
2	218	246	4	30L	8	2	547	549	26	146	-127	9	29L	1			18	28L	15	3	28L	44						
3	27L	44	5	30L	46	3	14L	-20	27	30L	-1	10	90	-89	2	468	596	19	195	-165	4	29L	-24					
4	63	80	6	29L	23	4	41	-47	28	52	69	11	81	-66	3	667	792	20	30L	40	5	53	-44					
5	106	-102	7	100	11	5	66	40	29	65	35	12	198	-188	4	295	323	21	31L	4	6	29L	-19					
6	83	-82	8	29L	21	6	15L	-30	80	-51	13	30L	-21	5	315	-274	22	117	99	7	118	83						
7	189	207	9	114	-14	87	7	107	114	31	72	73	14	169	173	6	158	-100	23	56	37	8	30L	22				
8	27L	-62	10	170	-184	8	121	-125	32	50	41	15	240	238	7	571	-509	24	75	-57	9	30L	-7					
9	28L	-52	11	128	-125	9	17L	17	33	27L	-24	16	30L	39	8	714	671	25	173	169	10	63	-69					
10	190	-191	12	28L	-19	10	132	-130	34	62	58	17	155	144	9	249	237	26	242	205	11	417	-379					
11	229	-253	13	65	67	11	82	-80	35	80	-99	18	119	-144	10	697	-673	27	34L	49	12	447	-438					
12	28L	-28	14	86	83	12	173	132	36	24L	25	19	106	-112	11	48	-5	28	34L	-19	13	277	-243					
13	51	76	15	127	139	13	113	-88	37	42	38	20	131	149	12	59	-72	29	197	-174	14	157	112					
14	92	91	16	91	89	14	194	-185	38	51	-67	21	14L	-147	13	22L	6	30	159	113	15	198	157					
15	29L	-6	17	81	-96	15	106	-118	39	19L	-28	22	7L	59	14	50	-44	31	34L	-21	16	748	692					
16	101	-42	18	44	56	16	125	-101	40	17L	-27	23	28L	-17	15	71	-147	32	33L	-17	17	33L	21					
17	30L	53	19	25L	28	17	275	-281				24	54	-70	16	417	-366	33	33L	20	18	126	-112					
18	78	-87	20	57	-39	18	96	85		3,5,L	25	27L	20	17	312	-293	34	32L	-10	19	131	107						
19	30L	24	21	41	126	19	220	214			26	26L	-15	18	7L	58	35	31L	-25	20	342	-306						
20	51	18	22	114	-129	20	26L	257	1	125	130	27	25L	13	19	211	123	36	30L	31	21	207	205					
21	31L	-26	23	108	47	21	102	-108	2	125	110	28	24L	25	20	320	-262	37	28L	-30	22	34L	10					
22	79	-31	24	41	17	22	262	258	3	117	106	29	42	50	21	357	250	38	119	-111	23	158	-144					
23	31L	35	25	19L	-103	23	14L	-122	4	69	71	30	22L	10	22	30L	38	39	46	-60	24	73	65					
24	59	-8	26	86	110	24	365	-341	5	108	-43	31	21L	10	23	120	-89	40	23L	22	25	34L	-19					
25	30L	-49	27	81	-21	25	306	308	6	354	-362		24	32L	21	41	51	26	26	34L	-30							
26	30L	-1					26	23Z	-218	7	356	-393		3,8,L	25	7L	45	42	78	-75								
27	55	29	2	10,L	27	29L	26	8	178	-148			26	7L	72	72	43	12L	78			4,6,L						
28	29L	35					28	14L	-126	9	122	72	1	30L	-46	27	34L	-32										
29	45	20	1	65	60	29	103	-97	10	256	24L	2	82	88	29L	-32		4,3,L		1	32L	49						
30	67	30	2	26L	-3	30	30L	33	11	166	166	3	165	-177	29L	-34L	14			2	129	-119						
31	27L	-48	3	26L	0	31	30L	-9	12	26L	32	4	116	137	30	34L	-22	0	158	-164	3	32L	-7					
32	38	27	4	26L	23	32	93	69	13	26L	-35	5	30L	25	31	34L	5	1	32L	320	4	32L	-5					
33	24L	21	5	26L	4	33	29L	44	14	145	140	6	62	-72	32	34L	-17	2	384	376	5	32L	-17					
34	49	-0	6	57	-20	34	29L	3	15	197	196	7	78	93	33	33L	-25	3	229	206	6	32L	-27					
35	42	52	7	25L	25	35	28L	21	16	261	230	8	29L	18	18	106	96	4	178	155	7	32L	8					
36	44	-60	8	71	3	36	27L	1	17	55	46	9	29L	-62	35	32L	45	5	91	-54	8	208	-213					
							9	54	-70	37	26L	-28	18	84	-66	10	29L	-30	36	31	-17	6	390	-352	9	120	-119	
10	24L	-25	38	59	-58	19	103	-89	11	29L	-37	4	37	54	-65	7	114	142	10	173	160							
11	24L	32	33	39	104	-99	20	29L	34	12	77	-63	38	135	-107	8	204	-163	11	125	268							
12	139	170	13	23L	13	40	22L	-39	21	29L	36	13	29L	-16	39	77	-86	9	235	-218	12	33L	61					
13	95	86	14	22L	15	42	54	46	23	46	-41	15	150	148	41	55	37	11	25L	-5	14	34L	32					
14	5	97	16	39	-35	44	11L	8	25	93	-86	17	93	93	43	111	138	13	146	-127	16	389	316					
15	30L	-4	17	50	43		26	50	88	18	27L	-36				14	27L	-9	17	34L	30							
16	230	27	17L	-21	15	120	124			1	28L	4	14	90	-113	31	33L	-37	34	21L	-7							
17	29L	18	8	17L	-21	15	120	124			2	28L	-13	15	71	54	32	33L	-41									
18	67	41	9	17L	7	16	140	131	1	52	44	2	28L	-13	15	71	54	32	33L	-41								
19	28L	-20	10	47	-75	17	100	-110	2	71	55	3	95	81	16	59	40	33	32L	26								
20	28L	-17	11	51	-81	18	93	-73	3	119	-119	17	4	100	-111	17	58	-62	34	55	40							
21	27L	-1					19	46	20	4	110	121	5	5	28L	-33	18	43	-45	35	84	-76	0	90	-86			
22	64	-60		3,1,L	20	24Z	-222	5	109	77	6	91	108	19	28L	26	36	28L	-51	1	34L	48						
23	47	-62					21	214	167	6	99	73	7	76	-59	20	62	94	37	27L	-0	2	112	148				
24	30L	-9	1	94	-69	22	248	-241	7	401	426	8	27L	59	21	156	144	38	43	53	3	34L	67					
25	23L	14	2	251	-293	23	28L	24	8	91	78	9	27L	-52	22	112	93	39	23L	-29	4	34L	8					
26	3	194	247	24	213	218	9	118	-127	10	27L	-40	23	31L	-38	40	20L	11	5	34L	.2							
27	4	277	-308	25	220	-210	17	275	-283	11	23L	-16	24	131	-95				6	34L	-56							
28	5	66	-73	26	135	151	11	278																				

TABLE III (continued).

4,7,L	32	32L -17	21	279 -272	16	134 -121	18	147	130	10	329 -311	10	63	43	15	29L -14			
	33	31L 14	22	113 -99	17	110 -164	19	99	93	11	314 -308	11	77	90	16	30L -44			
30	65	54	34	30L 15	23	34L 18	12	128 -115	20	97 -102	12	94	97	12	74	-85			
31	21L	27	35	29L -0	24	34L -5	19	120 -21	21	127 -152	13	159 -143	13	141 -152	18	158 -152			
32	33	-45	36	47	41	25	33L -55	20	31L -51	22	145 -151	14	146 -126	14	131 -144	19	31L -45		
33	16L	15	37	88	75	26	33L -34	21	30L -28	23	88 -64	15	69 -62	15	124 -122	20	31L -54		
34	11L	-32	38	170 -151	27	75	-55	22	29L -34	24	33L -27	16	32L -53	16	81 -65	21	31L -13		
	39	23L	9	28	32L	-28	23	60	-57	25	139 -121	17	32L -52	17	104 -91	22	144 -120		
	40	20L	17	29	32L	60	24	27L -21	26	69 -47	18	33L -1	18	47 -75	23	31L -36			
	41	71	-74	30	103	98	25	26L -2	27	33L -0	19	83 -62	19	28L -59	24	31L -43			
1	156	-156						28	110	78	20	76	-82	20	28L -38				
2	34L	87	5,2,L	32	29L	34	5,8,L	29	32L -12	21	33L -27	21	44 -48		7,3,L				
3	101	-102						30	62	-67	22	33L -28	22	44 -38					
4	34L	-9	1	79	-60	34	26L -8	1	105	101	31	31L -41	23	100 -60	23	58 -45			
5	81	2	46	25	23	24L -8	2	197	-178	32	30L -3	24	91 -81	24	24L -22	2	25L -19		
6	114	-115	3	18L	83			4	101	78		25	32L -18	25	22L -29	3	25L -11		
7	34L	-34	4	18L	-23	5,5,L	3	33L -23		6,2,L	26	31L -30		4	25L -41				
8	34L	-40	5	281	-276			5	32L -38		27	62	66	6,8,L	5	25L -14			
9	33L	-14	6	221	218	1	59	61	6	82	-83	0	181 -170	28	101 -71	6	152 -148		
10	33L	37	7	136	-140	2	29L -24	7	32L -43	1	174	170	29	29L -39	0	83 -87			
11	138	138	8	81	75	3	74	-82	8	64	73	2	43	23	30	28L -2			
12	72	87	9	132	105	4	81	-80	9	32L -43	3	101	-93	31	27L -2	2	30L -4		
13	81	71	10	263	-237	5	96	92	10	32L -19	4	20L	27	32	25L -19	3	63 -63		
14	32L	25	11	74	79	6	30L -16	11	31L -43	5	51	-23	33	24L -22	4	30L -55			
15	152	-161	12	124	141	7	30L -30	12	86 -89	6	171	192	34	22L -9	5	30L -52			
16	94	106	13	216	194	8	79	-88	13	80	-91	7	157 -141	35	19L -83	6	30L -25		
17	90	-85	14	42	-38	9	31L -15	14	30L -43	8	102 -122	36	61 -58	7	30L -13	14	60 -54		
18	30L	2	15	83	-61	10	188	168	15	79	69	24	25L -35	37	8	30L -16	15	54 -47	
19	81	71	16	141	136	11	283 -252	16	29L -53	10	62	-47		9	29L -24	16	70 -44		
20	110	-124	17	224	229	12	129	113	17	28L -83	11	110	115	6,5,L	10	29L -4	17	80 -35	
21	28L	1	18	159	148	13	32L -6	18	62	81	12	142	131	11	29L -30	18	68 -76		
22	27L	-12	19	102	74	14	211 -174	19	95 -109	13	68 -68	0	271 -296	12	28L -29	19	31L -37		
23	26L	15	20	349	-319	15	169	151	20	26L -6	14	117	129	1	73 -72	20	83 -73		
24	25L	-18	21	244	-211	16	33L -2	21	25L -26	15	59	12	2	31L -45	14	107 -97	21	31L -13	
25	24L	-43	22	81	-76	17	33L -33		16	95	96	3	31L -42	15	116 -119	22	83 -79		
26	79	-65	23	51	-58	18	33L -0		5,9,L	17	114 -133	4	31L -14	16	65 -84	23	31L -35		
27	21L	-25	24	156	138	19	159	124	18	120 -131	5	238 -237	17	59 -78	24	31L -20			
28	19L	40	25	69	-75	20	272 -252	1	29L -56	19	175 -115	6	102 -97	18	60 -63	25	48 -54		
29	85	109	26	34L	36	21	101	-89	2	193 -147	20	32L -5	7	69 -53	19	100 -114	26	74 -73	
30	13L	7	27	34L	46	22	33L -20	3	29L -38	21	237 -233	8	32L -28		27	29L -1			
	28	88	-86	23	114 -106	4	29L -7	22	352 -341	9	161 -165		6,9,L						
4,9,L	29	-100	98	24	33L	77	5	28L -47	23	64	-51	10	151 -122		7,4,L				
	30	33L	-2	25	93	96	6	28L -17	24	88 -69	11	174	151	0	72 -82				
0	94	-63	31	32L	-1	26	80	88	7	28L -0	25	203 -180	12	79 -66	1	48 -30	1	164 -176	
1	31L	41												2	28L -51	2	28L -16		
2	99	81	5,3,L	27	31L	49	8	28L -27	26	169	145	13	133 -108	2	25L -35	3	83 -83		
3	31L	78		28	30L	10	9	27L -32	27	33L -28	14	262 -258	3	25L -35					
4	76	56	1	95	90	30	28L -24	11	26L -18	19	32L -21	15	68 -60	4	25L -15	4	201 -227		
5	31L	-70	3	68	-48	32	86	77	13	25L -61	31	30L -30	18	81 -88	7	24L -3	7	124 -108	
7	31L	53	4	22L	-23	33	55 -39	14	25L -28	28	29L -21	16	19	81	73	8	24L -3	8	139 -126
8	30L	-52	5	244	231	34	50 -52	15	24L -18	33	81 -53	20	85 -72	9	84 -16	9	207 -181		
9	30L	20	6	153	-147	35	20L -15	16	61 -87	34	64 -48	21	32L -41	10	122 -43	10	137 -132		
10	30L	2	7	64	-70	36	17L -30	17	54 -76	35	25L -26	22	62 -40	11	22L -40	11	30L -18		
11	29L	-32	8	128	-145			18	68 -65	36	23L -33	23	32L -48	12	69 -80	12	80 -78		
12	29L	-55	9	81	-56	5,6,L	19	47	36		24	56 -25	13	125 -122	13	67 -66			
13	28L	-6	10	307	281			20	19L -3		6,3,L	25	30L -7	14	20L -22	14	109 -74		
14	28L	21	11	350	-330	1	32L -38	21	17L -37		26	29L -37	15	19L -54	15	75 -58			
15	27L	-32	12	27L	15	2	109 -120		0	139	141	27	80 -56		16	31L -29			
16	67	68	13	27L	-10	3	72 -112	6,0,L	1	63	-38	28	63 -27		7,1,L	17	74 -76		
17	82	-71	14	147	-123	4	55 -51		2	39	32	29	26L -39		18	31L -34			
18	48	-50	15	154	160	5	33L -20	4	90 -168	3	71 -64	30	25L -35	1	141 -179	19	31L -21		
19	65	43	16	175	167	6	33L -53	5	220 -246	4	54 -36	31	47 -46	2	64 -74	20	31L -36		
20	50	16	17	268	235	7	33L -34	6	176 -193	5	143 -134	32	83 -74	3	90 -70	21	47 -24		
21	22L	16	18	264	234	8	66 -69	7	34 -11	6	133 -112	33	49 -40	4	118 -136	22	31L -40		
22	20L	-3	19	72	41	9	33L -10	8	38 -10	7	25L -67	34	161 -44	5	160 -199	23	53 -31		
23	30	329	-292	10	33L	54	9	152 -102	8	114 -98	35	11L -53	6	148 -198	24	43 -45			
24	22	99	-85	12	149	134	11	22L -103	10	208 -201		6,6,L	8	237 -258	26	63 -25			
25	21	267	-280	11	187	-151	10	86	53	9	237 -182		7	20L -22	25	45 -45			
26	22	99	-85	12	149	134	11	22L -103	10	208 -201		6,6,L	8	237 -258	26	63 -25			
27	1	91	-117	23	33L	25	13	45	12	283 -290	12	118 -147	17	9	454 -477	27	27L -17		
2	26L	74	24	165	136	14	233 -201	13	181 -191	12	192 -152	0	158 -163	10	156 -126	28	26L -35		
3	91	-129	25	34L	18	15	266	208	14	124 -124	13	363 -326	1	131 -30	29	45 -19			
4	25L	-26	10	26	65	16	83 -67	15	214 -201	14	216 -189	2	33L -20	12	25L -23	30	23L -21		
5	72	73	27	23L	53	17	183	185	16	28L -22	15	141 -123	3	120 -139	15	26L -9	31	62 -33	
6	25L	-79	28	33L	-16	18	222	217	17	184 -145	16	64 -20	4	33L -47	14	52 -47	32	42 -59	
7	24L	41	29	33L	-35	19	234 -244	18	156 -149	17	31L -17	5	88 -90	15	46 -18	33	37 -34		
8	50	78	-77	20	88	19	63	52	18	151 -119	6	147 -156	16	49 -26	34	12L -4			
9	51	-24	21	80	-104	20	110	109	19	32L -44	7	33L -21	17	72 -63					
10	125	120	33	29L	-5	23	97	-95		31	29L	43	19	85 -89	29	29L -14	10	85 -43	
11	101	-111	34	46	-26	24	160	-139	23	33L -65	23	33L -25	11	61 -45	21	76 -86	2	31L -13	
12	27	-38	35	27L	-2	25	30L	16	24	33L -49	25	33L -13	12	33L -18	22	31L -26	3	80 -89	
13	510	317	36	45	27	28L	-42	26	220	198	25	100	13	33L -26	23	355 -346	4	100 -110	
14	6	319	-352	30	66</														

TABLE III (*continued*).

7,6,L	16	30L	14	22	26L	14	28	50	45	2	30L	29	8	29L	19	14	27L	13	20	23L	-3
12	31L	-27	16	59	-31	24	24L	-8		3	30L	44	9	101	92	15	62	-44	21	43	-28
13	70	-57	19	28L	9	25	43	45	7,7,L	5	85	-54	11	29L	1	17	26L	-14	22	20L	0
14	30L	32	20	50	-21	26	21L	3		6	30L	-32	12	28L	17	18	42	35		19L	-3
15	30L	-6	21	27L	3	27	59	-70	1	100	-99	7	30L	16	13	58	-55	19	64	66	

TABLE IV.

*Interatomic angles.*

apex	end	end		apex	end	end	
<i>sp<sup>3</sup> atoms</i>							
C <sub>1</sub>	C <sub>2</sub>	C <sub>11</sub>	110°.81	C <sub>2</sub>	C <sub>1</sub>	C <sub>3</sub>	112°.20
C <sub>1</sub>	C <sub>2</sub>	C <sub>15</sub>	109°.86	C <sub>3</sub>	C <sub>2</sub>	C <sub>4</sub>	110°.67
C <sub>1</sub>	C <sub>2</sub>	C <sub>16</sub>	106°.77	C <sub>4</sub>	C <sub>3</sub>	C <sub>12</sub>	111°.47
C <sub>1</sub>	C <sub>11</sub>	C <sub>15</sub>	109°.71	C <sub>17</sub>	C <sub>7</sub>	C <sub>18</sub>	111°.95
C <sub>1</sub>	C <sub>11</sub>	C <sub>16</sub>	110°.78	C <sub>17</sub>	C <sub>7</sub>	C <sub>19</sub>	111°.63
C <sub>1</sub>	C <sub>15</sub>	C <sub>16</sub>	108°.84	C <sub>17</sub>	C <sub>18</sub>	C <sub>19</sub>	111°.10
<i>sp<sup>2</sup> atoms</i>							
C <sub>5</sub>	C <sub>6</sub>	C <sub>13</sub>	122°.87	C <sub>11</sub>	C <sub>10</sub>	C <sub>12</sub>	119°.69
C <sub>6</sub>	C <sub>5</sub>	C <sub>7</sub>	119°.99	C <sub>12</sub>	C <sub>4</sub>	C <sub>11</sub>	121°.64
C <sub>7</sub>	C <sub>6</sub>	C <sub>8</sub>	118°.43	C <sub>12</sub>	C <sub>4</sub>	C <sub>13</sub>	117°.82
C <sub>7</sub>	C <sub>6</sub>	C <sub>17</sub>	120°.06	C <sub>12</sub>	C <sub>11</sub>	C <sub>13</sub>	120°.52
C <sub>7</sub>	C <sub>8</sub>	C <sub>17</sub>	121°.51	C <sub>13</sub>	C <sub>5</sub>	C <sub>12</sub>	123°.87
C <sub>8</sub>	C <sub>7</sub>	C <sub>14</sub>	122°.57	C <sub>13</sub>	C <sub>5</sub>	C <sub>14</sub>	117°.33
C <sub>9</sub>	C <sub>10</sub>	C <sub>14</sub>	120°.46	C <sub>13</sub>	C <sub>12</sub>	C <sub>14</sub>	118°.80
C <sub>10</sub>	C <sub>9</sub>	C <sub>11</sub>	120°.74	C <sub>14</sub>	C <sub>8</sub>	C <sub>9</sub>	121°.48
C <sub>11</sub>	C <sub>1</sub>	C <sub>10</sub>	116°.31	C <sub>14</sub>	C <sub>8</sub>	C <sub>13</sub>	118°.76
C <sub>11</sub>	C <sub>1</sub>	C <sub>12</sub>	124°.01	C <sub>14</sub>	C <sub>9</sub>	C <sub>13</sub>	119°.75

TABLE V.

PLANE FORMED BY	EQUATIONS FOR		
	Direct space	Angstrom space	
$13\ sp^2$ bonded atoms . . .	(1D) $4.537x + 7.222y + 13.168z = 7.029$	(1A) $0.491x + 0.790y + 0.366z = 7.029$	
all C atoms . . .	(2D) $4.298x + 7.575y + 11.147z = 6.867$	(2A) $0.465x + 0.829y + 0.310z = 6.867$	
DISTANCES IN ANGSTROM FROM THE PLANE (1 A)			
C <sub>1</sub> . . . . .	0.031	C <sub>10</sub> . . . . .	-0.001
C <sub>4</sub> . . . . .	-0.074	C <sub>11</sub> . . . . .	0.004
C <sub>5</sub> . . . . .	0.051	C <sub>12</sub> . . . . .	-0.015
C <sub>6</sub> . . . . .	0.049	C <sub>13</sub> . . . . .	0.010
C <sub>7</sub> . . . . .	-0.002	C <sub>14</sub> . . . . .	0.003
C <sub>8</sub> . . . . .	-0.022	C <sub>17</sub> . . . . .	-0.036
C <sub>9</sub> . . . . .	0.000	$\sigma$ . . . . .	0.033

TABLE VI.  
*Intermolecular distances smaller than 4 Å*

Atom 1	Atom 2	Symmetry operation on atom 2	Distances in Å
C <sub>6</sub>	C <sub>16</sub>	$1+x \quad y \quad z$	3.78
C <sub>18</sub>	C <sub>10</sub>	$1+x \quad y \quad z$	3.81
C <sub>17</sub>	C <sub>19</sub>	$3/2-x \quad 1/2-y \quad z$	3.88
C <sub>15</sub>	C <sub>15</sub>	$x \quad 3/2-y \quad 1/2-z$	3.97
C <sub>4</sub>	C <sub>15</sub>	$1/2+x \quad 3/2-y \quad z$	3.91
C <sub>8</sub>	C <sub>10</sub>	$1/2+x \quad 3/2-y \quad z$	3.95
C <sub>12</sub>	C <sub>15</sub>	$1/2+x \quad 3/2-y \quad z$	3.76
C <sub>13</sub>	C <sub>15</sub>	$1/2+x \quad 3/2-y \quad z$	3.89
C <sub>15</sub>	C <sub>16</sub>	$1/2+x \quad 3/2-y \quad z$	3.95
C <sub>3</sub>	C <sub>2</sub>	$1/2+x \quad y \quad 1/2-z$	3.87
C <sub>6</sub>	C <sub>12</sub>	$1/2+x \quad 1/2-y \quad z$	3.94
C <sub>6</sub>	C <sub>13</sub>	$1/2+x \quad 1/2-y \quad z$	3.96
C <sub>19</sub>	C <sub>8</sub>	$1/2+x \quad 1/2-y \quad z$	3.98

The large angle at C<sub>13</sub> (ends C<sub>5</sub> and C<sub>12</sub>) and perhaps also the long distance C<sub>12</sub>-C<sub>13</sub> are probably caused by interactions of this kind. The effect is less comprehensible here, as C<sub>4</sub> is a tetrahedrally bonded atom, and its two hydrogens (instead of a single one) ought to be symmetrically disposed in respect of the least square plane 1A and then of H<sub>5</sub>. The observed distances between the hydrogens attached to C<sub>4</sub> and H<sub>5</sub> are in fact 2.1 Å.

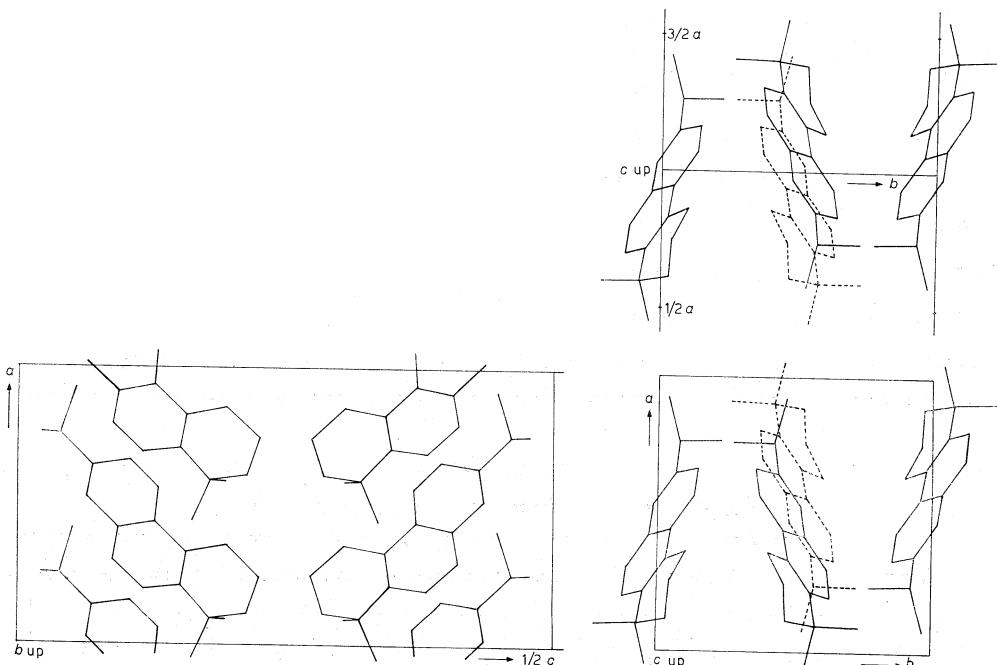


Fig. 4. - Projections of the simonellite structure along [010] ( $\alpha$ , half cell) and [001] ( $\beta$ ); the second projection is divided into two parts, to overcome exaggerated superposition.

The molecular packing is shown in fig. 4: fig. 4  $\beta$  has been split owing to the overlapping of the molecules along the  $c$  axis. Few intermolecular contacts smaller than 4 Å are shown in Table VI.

It is interesting to notice how the molecules lie almost linearly along  $\alpha$ , in agreement with high refraction index in this direction, called  $b$  by Emliani and Pellizzer (1952).

The possible biogenetical provenience of simonellite could be a double-step action on abietic acid (fig. 1 c): hydrogenation of the carboxyl group and contemporary loss of the methyl group at C<sub>12</sub>, followed by an incomplete aromatization.

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